



Health & Welfare

Feed ingredient digestibility

Monday, 1 December 2003

By Stephen Goddard, Ph.D., Saiyed Ahmed, Ph.D. and Ghazi Al-Shagaa

Oman lab refines in vitro testing tor tilapia feed





The Agricultural Experiment Station at Sultan Qaboos University is applying modified test methods to evaluate potential ingredients for tilapia feed.

Considerable research has been conducted with tilapia to evaluate cost-effective and regionally available sources of protein for inclusion in tilapia feed. However, fishmeal and soybean meal remain the protein sources of choice for feed manufacturers.

Fishmeal has the correct balance of essential amino acids, is highly palatable and digestible, and has additional properties as an appetite stimulant or attractant. Typical inclusion levels in complete tilapia feeds for use in intensive culture are in the 10-15 percent range of the total diet.

Many alternate protein sources of animal and plant origin continue to be investigated. However, for various reasons, most of them can not completely replace fishmeal. Some ingredients lack sufficient essential nutrients, others contain toxic substances or antinutritional factors, while others reduce palatability.

Measuring digestibility: in vivo methods

Determining the digestibility of protein and other nutrients in feed ingredients is an important step in the formulation of balanced feed. Digestibility, a measure of the quantity of ingested nutrients retained by fish, is most commonly measured by indirect methods using inert marker materials.

By adding an inert material to the feed (external marker) or measuring an inert natural component of the food (internal marker), the apparent digestibility (AD) can be calculated by comparing the ratio of the marker in the food and feces to a specific nutrient. In the formula for determining AD below, the term apparent digestibility is used since feces are contaminated with endogenous material, and hence the values are not "true" values.

AD (percent) = $[1 - (F/D \times Dm/Fm)] \times 100$

Where AD = apparent digestibility F = percent nutrient or energy in feces D = percent nutrient or energy in diet Dm = percent marker in diet Fm = percent marker in feces.

To be effective, markers must be indigestible, nontoxic, and completely inert. They should move through the gut at the same rate as the digesta.

By far the most commonly used external marker in digestibility studies, chromic oxide is also widely used in tilapia studies. Some results, however, question its validity with tilapia. There is some evidence that chromic oxide can partially separate from food during ingestion and passage through the gut, and that it may not be totally inert. Previous reports from the authors' laboratory suggest the use of acid-insoluble ash as an alternative marker for tilapia studies.

Digestibility values for individual ingredients are determined by substitution of the test ingredient in a reference diet. Digestibility values for both reference and test diets are measured and values substituted in the following formula:

AD of test ingredient = AD of test diet - 0.7 AD of reference diet/0.3 AD of test diet

There is little standardization of methodology for feeding and the collection of fecal material during *in vivo* digestibility studies. Rapid separation of feces from the water is essential to reduce the contamination, breakup, or loss of soluble materials. This is generally performed using a trap or conical tank from which feces can be removed periodically through a collection arm.

Measuring digestibility: in vitro methods

The published digestibility values of ingredients used to formulate tilapia feed were obtained using the *in vivo* methods outlined above. Measurement of ingredient digestibility using *in vitro* methods may provide an additional measure of the biological value of proteins for use in aquafeed. Various approaches have been taken, including the use of single or multiple enzyme treatments to simulate peptic digestion, monitoring pH changes, and maintenance of pH using pH-stat methods to record the hydrolysis of peptide bonds.

Current studies in the College of Agricultural and Marine Sciences at Sultan Qaboos University in Muscat, Oman, focus on the measurement of free amino groups released by tilapia enzymes from ingredient samples. Modified from *in vitro* digestibility studies with salmonids, the study method has been used to compare fishmeal and fish hydrolysates prepared from fisheries by-catch for use in tilapia feed.

In this method, enzymes are extracted from homogenized proximal intestine and separated by centrifugation and dialysis. Chymotrypsin is then assayed against an N-Succinyl-LAla-LAla-LPro-LPhe-pNitroanilide substrate to determine activity levels.

The required 1 DA per minute enzyme volume is added to ingredient samples, which are stirred and incubated at 25 degrees-C for 18 hours. Samples of 1 ml are then cooled and centrifuged. Trinitrobenzensulphonic acid (TNBS) at 0.1 percent is added to each sample, which is then incubated for one hour at 60 degrees-C. When the samples cool, the reaction is stopped by the addition of acid and absorbance measured at 420 nm.

The TNBS measurements are transformed into alanine equivalents from a DL alanine standard curve. As an example,

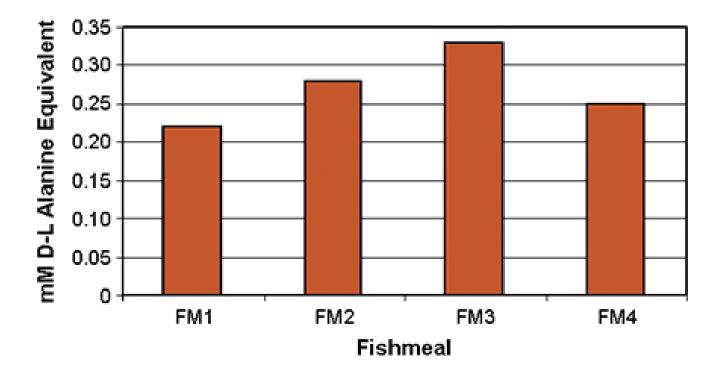


Fig. 1: Enzyme-released free amino groups in alanine (mM) equivalents from samples prepared from fisheries by-catch.

the values of enzyme-released, free amino groups from four fishmeal samples are shown in Fig. 1.

The test detects differences in the quantity of amino acids released by enzymes from protein sources. The initial results obtained from *in vitro* measurements were closely correlated with the amino acid profiles of the tested fishmeal and results obtained from tilapia growth experiments.

Conclusion

Test results confirmed that *in vitro* enzyme analyses can be utilized for the assessment of the nutritive value of aquafeed ingredients. Researchers are now extending their work to include a wider range of protein sources and the use of more purified and optimized enzyme fractions.

(Editor's Note: This article was originally published in the December 2003 print edition of the Global Aquaculture Advocate.*)*

Authors



STEPHEN GODDARD, PH.D.

Department of Marine Science and Fisheries College of Agricultural and Marine Sciences Sultan Qaboos University P.O. Box 34, Al-Khod 123 Muscat, Oman

sgoddard@squ.edu.om (mailto:sgoddard@squ.edu.om)



SAIYED AHMED, PH.D.

Department of Marine Science and Fisheries College of Agricultural and Marine Sciences Sultan Qaboos University P.O. Box 34, Al-Khod 123 Muscat, Oman



GHAZI AL-SHAGAA

Department of Marine Science and Fisheries College of Agricultural and Marine Sciences Sultan Qaboos University P.O. Box 34, Al-Khod 123 Muscat, Oman

Copyright © 2016–2020 Global Aquaculture Alliance

All rights reserved.